



**Aprovecho Research Center**

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PARTNERSHIP FOR CLEAN INDOOR AIR (PCIA)



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# Stove Design and Performance Training

**Dhaka Bangladesh**  
**November 30-December 2 2011**

**Mike Hatfield – Aprovecho Reserch Center, USA**



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# Aprovecho Research Center

Started in 1976 with invention  
of Lorena Stove

A failure!!!!

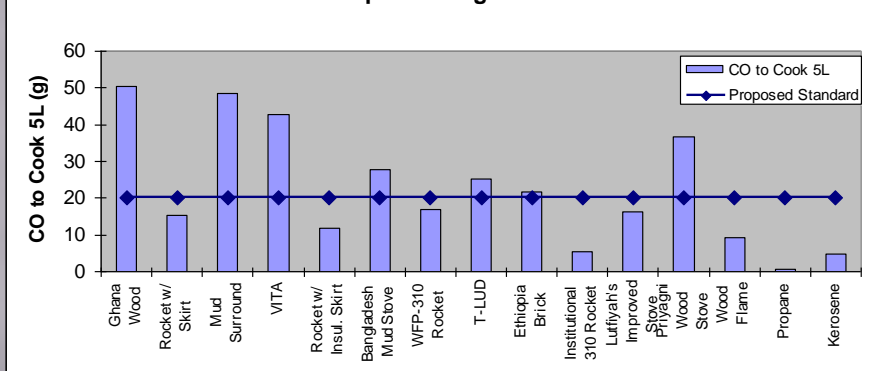




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CO to Cook 5L vs. Proposed 20 g Standard of Performance



Two focuses

1 - In Field Projects

2 - Monitoring and Evaluation

# Aprovecho Stove Projects and Trainings

Mexico  
Guatemala  
Honduras  
El Salvador  
Nicaragua

In Field Projects

Ghana

Brazil  
Bolivia

Uganda  
Malawi  
South Africa  
Lesotho  
Zaire

Indonesia

India  
Philippines  
Tibet  
China

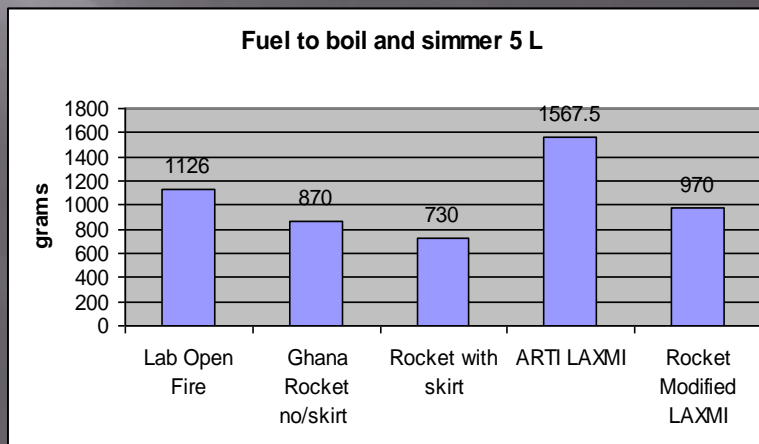
Active Stove Projects in over 20 countries – Trainings in 30 countries with participants from over 60 countries – Tested and evaluated over 100 stoves



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## Lab Work



International Testing Center

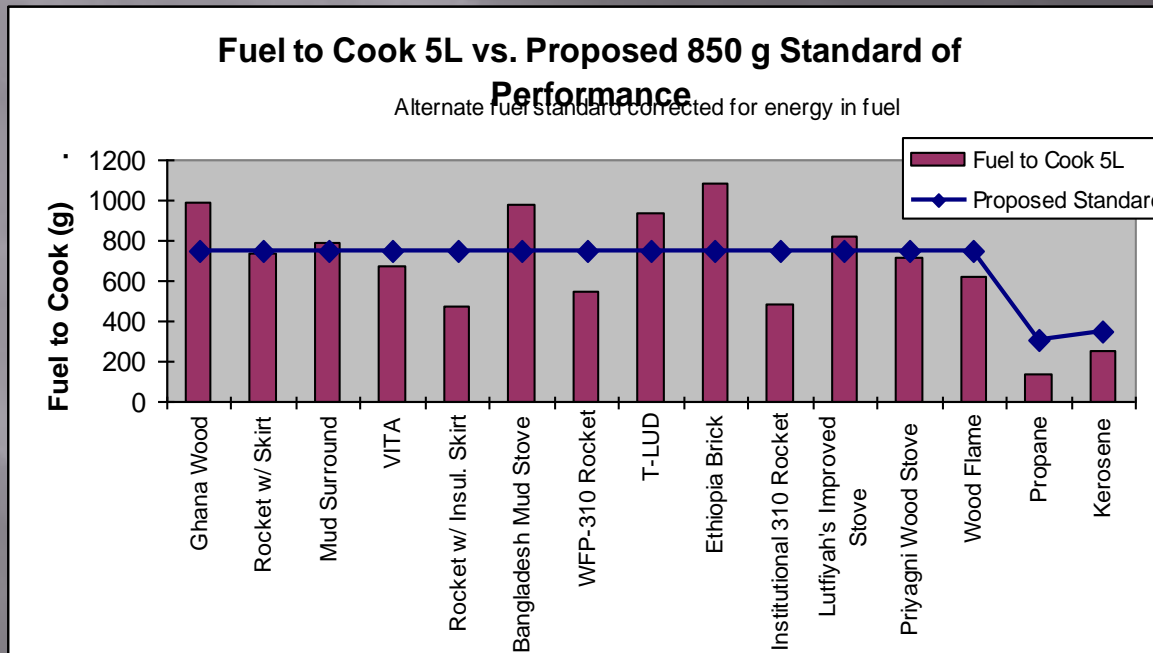
Creation of Regional Testing Centers



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## Lab Work



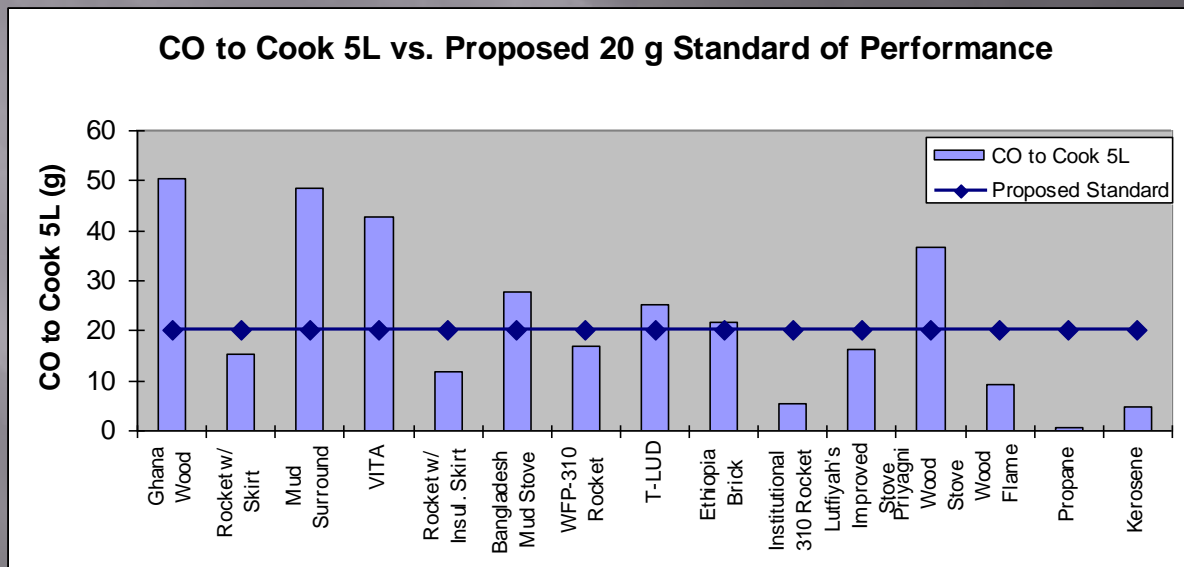
Creating International Standards of Performance



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## Lab Work



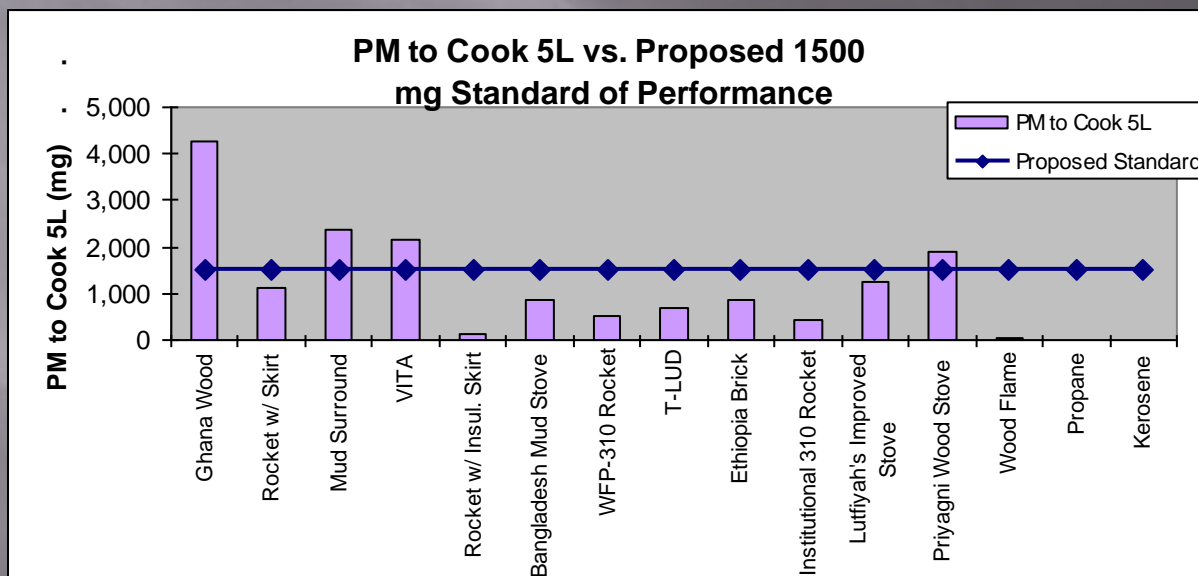
Creating International Standards of Performance



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## Lab Work



Creating International Standards of Performance





# Elements of an improved stove (ICS)

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There are four goals we need to meet when designing a stove —

- 1- The stove cooks food as well or better than the traditional method
- 2- The stove eliminates or reduces the amount of smoke in the kitchen
- 3- It uses less fuel to cook food
- 4- Is producible at a cost that is acceptable to users



# Elements of an improved stove

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- 1 - The stove cooks food as well or better than the traditional method
  - ▣ Use local cooks throughout design process
  - ▣ Form a stove committee
  - ▣ Perform tests using local cooks (CCT and KPT)
  - ▣ Follow up by independent organizations



# Elements of an improved stove

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2 - The stove eliminates or reduces the amount of smoke in the kitchen

- ▣ Worldwide 1.6 million people, mostly women and young children, die each year from breathing wood smoke!!
- ▣ By cleaning up combustion as best we can and then making sure cooks are not exposed to what smoke is produced we reduce health risks of biomass cooking.
- ▣ Chimneys



## Elements of an improved stove

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3 - It uses less fuel to cook food

- ▣ Often meeting goals one and two will be in direct conflict with fuel savings
- ▣ Requires testing (WBT, CCT, KPT) to determine if fuel is in truth being saved



# Elements of an improved stove

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4 - Is producible at a cost that is acceptable to users

- ❑ Given enough money almost any stove can be made
- ❑ To reach the majority of the some 3 billion people who cook on biomass we need to have a stove that sells for as little as \$5
- ❑ Other options – Micro Finance, Carbon Credits...?



# Simplified Designing of an Improved Stove

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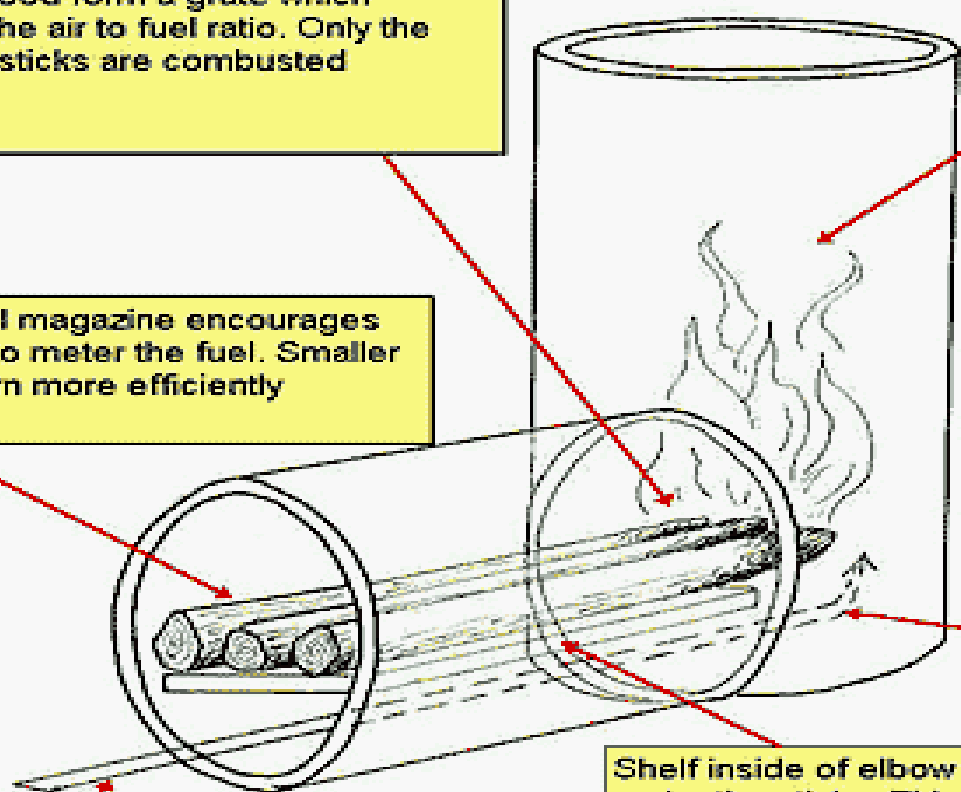
- 1 - Improved Combustion and/or chimney
- 2 - Improved Heat Transfer

# There are many ways to achieve our goals

sticks of wood form a grate which improves the air to fuel ratio. Only the tips of the sticks are combusted

Small fuel magazine encourages the user to meter the fuel. Smaller sticks burn more efficiently

Rocket chimney increases draft. Smoke is drawn through flame and combusts



Air passing under the shelf is preheated

Shelf inside of elbow allows air to pass under the sticks. This ensures optimal airflow into combustion chamber

Fuel magazine helps to limit the inflow of cool air. Cool air reduces the temperature in the the combustion chamber and decreases efficiency.



# Simplified combustion theory review

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- ❑ Wood doesn't burn
- ❑ Wood gets hot and releases volatile gases that then combust
- ❑ If wood is heated to 650 degrees Celsius (and sufficient oxygen is mixed with the volatile gases) the result is complete combustion. The products of clean combustion are CO<sub>2</sub>, water vapour and heat.
- ❑ For this to happen we need to have sufficient **Time, Temperature and Turbulence**
- ❑ A lot of heat, roughly speaking, dry wood has **half** the energy per kg as gasoline
- ❑ Smoke is wasted energy



# What are limiting factors to high temperatures and achieving complete combustion ?



## Challenge # 1

- ▣ **Cool stove body**
- ▣ **Cool earth**
- ▣ The body of the stove or of the earth **robs** heat from the fire
- ▣ Which lowers combustion temperature, decreases efficiency, and increases smoke



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# Lorena Stove

1 m x 1 m x 1 m = 1,000,000 cc

Density of 1.5 g/cc – 1,500 KG

-Tunnels = about 1,000 KG

1 kg clay raised 1 deg C = 1.38 KJ

1.38 KJ x 100 deg x 1,000 KG

= 138 MJ or 7KG dry wood

Plus thermal conductivity losses



# What are limiting factors to high temperatures and achieving complete combustion ?



## Challenge # 1

- ❑ **Cool stove body**
- ❑ **Cool earth**
- ❑ The body of the stove or of the earth **robs** heat from the fire
- ❑ Which lowers combustion temperature, decreases efficiency, and increases smoke

## Solution?

- ❑ **Insulate the stove** with low mass, heat resistant materials in order to keep the fire as hot as possible
- ❑ Remember mass is the opposite of insulation
- ❑ Effective stove insulators are pumice, vermiculite, and wood ash
- ❑ Dense things such as earth, sand, cement, water and cast iron are poor insulators



# Maximizing combustion efficiency

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## ▣ Challenge #2

- ▣ Cool wood
- ▣ Which lowers combustion temperatures...which decreases efficiency...And increases smoke



# Maximizing combustion efficiency

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## ▣ Challenge #2

- ▣ Cool wood
- ▣ Which lowers combustion temperatures...which decreases efficiency...And increases smoke

## ▣ Solution?

- ▣ Meter the fuel!
- ▣ Use small sticks whenever possible
- ▣ Maximize the surface area of the wood exposed to coals
- ▣ Heat only the fuel that is burning
- ▣ Burn the tips of sticks only as they enter the combustion chamber



# Maximizing combustion efficiency

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- ▣ Challenge # 3
- ▣ Cool air/ Too much air
- ▣ Which lowers combustion temperature, decreases efficiency, and increases smoke
- ▣ Note: an open fire can draw 20 times more than is required for stoichiometric (chemically ideal) combustion



# Maximizing combustion efficiency

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## ▣ Challenge # 3

- ▣ Cool air/ Too much air
- ▣ Which lowers combustion temperature, decreases efficiency, and increases smoke
- ▣ Note: an open fire can draw 20 times more than is required for stoichiometric (chemically ideal) combustion

## ▣ Solution ?

- ▣ Do not allow too much or too little air to enter the combustion chamber.
- ▣ There should be a minimum excess of air supporting clean burning.



# Maximizing combustion efficiency

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## ▣ Challenge # 4

- ▣ Cool cooking pot
- ▣ The cooking pot is generally no more than a 100 -200 degrees Celsius
- ▣ Flames touching the pot?
- ▣ **Soot and smoke!**





# Maximizing combustion efficiency

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## ▣ Challenge # 4

- ▣ Cool cooking pot
- ▣ The cooking pot is generally no more than a 100 -200 degrees Celsius
- ▣ Flames touching the pot?
- ▣ **Soot and smoke!**

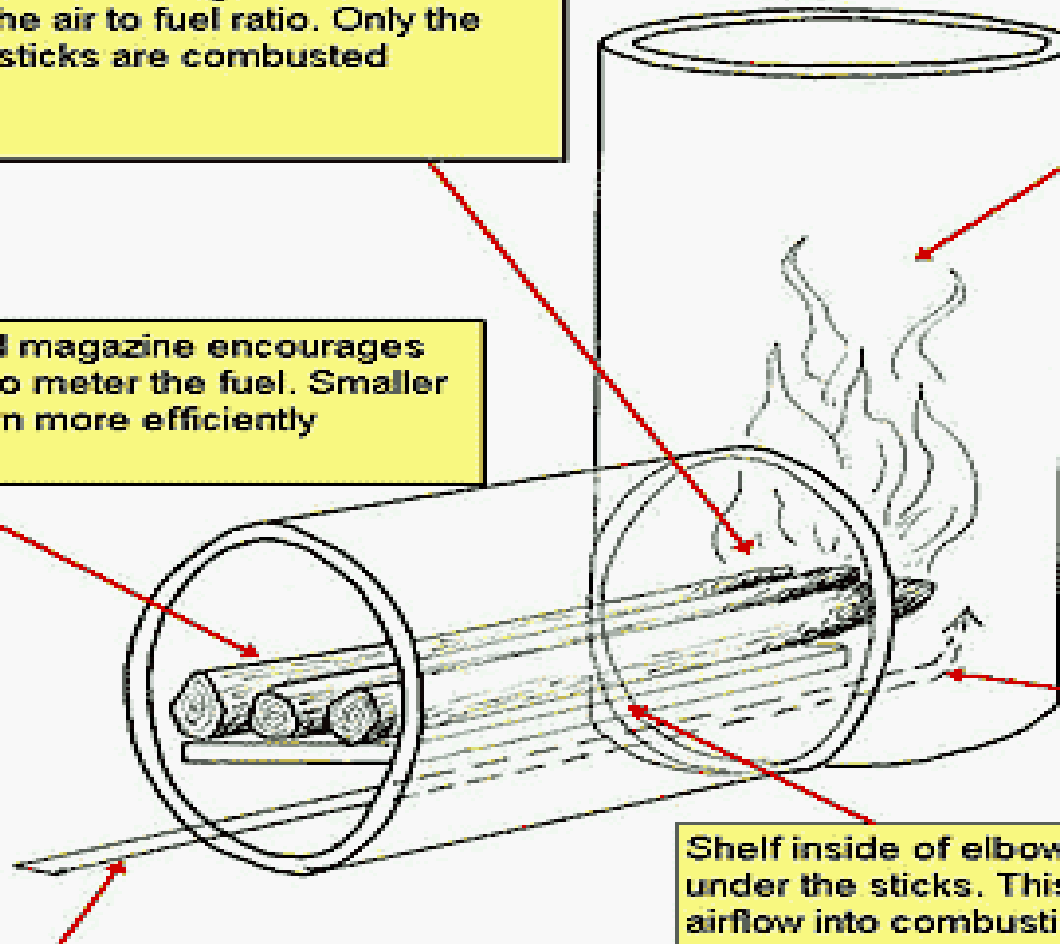
## Solution?

- ▣ Elevate the pot above the height of the flames
- ▣ This creates an internal 'chimney' which increases draft
- ▣ And gives time for improved air/ fuel mixing

Rocket chimney increases draft. Smoke is drawn through flame and combusts

sticks of wood form a grate which improves the air to fuel ratio. Only the tips of the sticks are combusted

Small fuel magazine encourages the user to meter the fuel. Smaller sticks burn more efficiently



Stove  
Wood  
Air  
Pot

Air passing under the shelf is preheated .

Shelf inside of elbow allows air to pass under the sticks. This ensures optimal airflow into combustion chamber

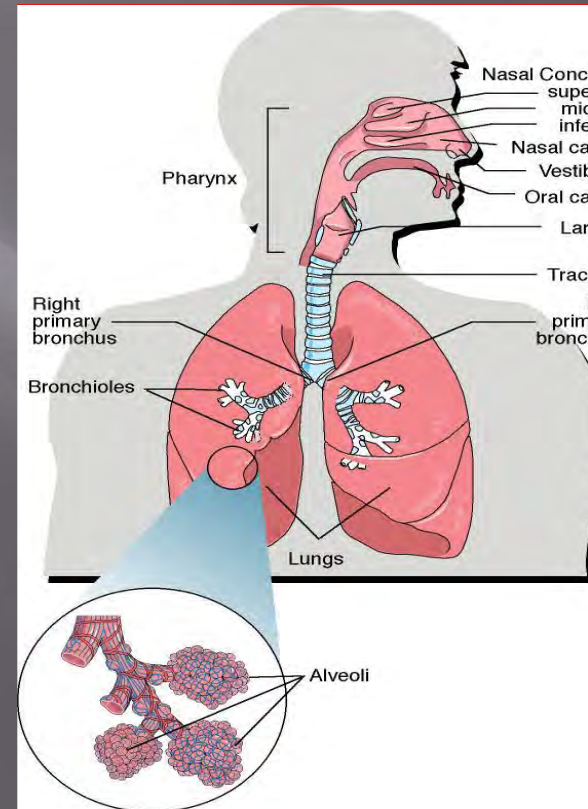
Fuel magazine helps to limit the inflow of cool air. Cool air reduces the temperature in the the combustion chamber and decreases efficiency.



# Complete combustion

Carbon Monoxide (CO) -  
Odorless and invisible

Particulate Matter (PM)  
- Visible Smoke



# A Few of the Chemicals in Woodsmoke (~g/kg emission factors)

Carbon Monoxide	80-370	Oxygenated PAHs	0.15-1
Methane	14-25	Polycyclic Aromatic Hydrocarbons (PAH)	
VOCs (C2-C7)	7-27	Fluorene	10 <sup>-5</sup> - 1.7x10 <sup>-2</sup>
Aldehydes	0.65.4	Phenanthrene	2x10 <sup>-5</sup> - 3.4x10 <sup>-2</sup>
Formaldehyde	0.1-0.7	Anthracene	5x10 <sup>-5</sup> - 2.1x10 <sup>-5</sup>
Acrolein	0.02-0.1	Methylanthracenes	7x10 <sup>-5</sup> - 8x10 <sup>-5</sup>
Propionaldehyde	0.1-0.3	Fluoranthene	7x10 <sup>-4</sup> - 4.2x10 <sup>-2</sup>
Butryaldehyde	0.01-1.7	Pyrene	8x10 <sup>-4</sup> - 3.1x10 <sup>-2</sup>
Acetaldehyde	0.03-0.6	Benzo(a)anthracene	4x10 <sup>-4</sup> - 2x10 <sup>-3</sup>
Furfural	0.2-1.6 1.6	Chrysene	5x10 <sup>-4</sup> - 1x10 <sup>-2</sup>
Substituted Furans	0.15-1.7	Benzofluoranthenes	6x10 <sup>-4</sup> - 5x10 <sup>-3</sup>
Benzene	0.6-4.0	Benzo(e)pyrene	2x10 <sup>-4</sup> - 4x10 <sup>-3</sup>
Alkyl Benzenes	1-6	Benzo(a)pyrene	3x10 <sup>-4</sup> - 5x10 <sup>-3</sup>
Toluene	0.15-1.0	Perylene	5x10 <sup>-5</sup> - 3x10 <sup>-3</sup>
Acetic Acid	1.8-2.4	Ideno(1,2,3-cd)pyrene	2x10 <sup>-4</sup> - 1.3x10 <sup>-2</sup>
Formic Acid	0.06-0.08	Benz(ghi)perylene	3x10 <sup>-5</sup> - 1.1x10 <sup>-2</sup>
Nitrogen Oxides (NO,NO <sub>2</sub> )	0.2-0.9	Coronene	8x10 <sup>-4</sup> - 3x10 <sup>-3</sup>
Sulfur Dioxide	0.16-0.24	Dibenzo(a,h)pyrene	3x10 <sup>-4</sup> - 1x10 <sup>-3</sup>
Methyl chloride	0.01-0.04	Retene	7x10 <sup>-3</sup> - 3x10 <sup>-2</sup>
Napthalene	0.24-1.6	Dibenz(a,h)anthracene	2x10 <sup>-5</sup> - 2x10 <sup>-3</sup>
Substituted Napthalenes	0.3-2.1	Trace Elements	
Oxygenated Monoaromatics	1 - 7	Cr	2x10 <sup>-5</sup> - 3x10 <sup>-3</sup>
Guaiacol (and derivatives)	0.4-1.6	Mn	7x10 <sup>-5</sup> - 4x10 <sup>-3</sup>
Phenol (and derivatives)	0.2-0.8	Fe	3x10 <sup>-4</sup> - 5x10 <sup>-3</sup>
Syringol (and derivatives)	0.7-2.7	Ni	1x10 <sup>-6</sup> - 1x10 <sup>-3</sup>
Catechol (and denvatives)	0.2-0.8	Cu	2x10 <sup>-4</sup> - 9x10 <sup>-4</sup>
Particulate Organic Carbon	2-20	Zn	7x10 <sup>-4</sup> - 8x10 <sup>-3</sup>
Chlorinated dioxins	1x10 <sup>-5</sup> - 4x10 <sup>-5</sup>	Br	7x10 <sup>-5</sup> - 9x10 <sup>-4</sup>
Particulate Acidity	7x10 <sup>-3</sup> - 7x10 <sup>-2</sup>	Pb	1x10 <sup>-4</sup> - 3x10 <sup>-3</sup>
Normal alkanes (C <sub>24</sub> -C <sub>30</sub> )	1x10 <sup>-3</sup> - 6x10 <sup>-3</sup>	Elemental Carbon	0.3 - 5
		Cyclic di-and triterpenoids	
		Dehydroabietic acid	0.01 - 0.05
		Isopimaric acid	0.02 - 0.10
		Lupenone	2x10 <sup>-3</sup> - 8x10 <sup>-3</sup>
		Friedelin	4x10 <sup>-6</sup> - 2x10 <sup>-5</sup>



# Optimising heat transfer

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Baldwin found that even a smoky fire can be as high as 92%

Combustion Efficiency



## The second half of the equation

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The true efficiency or fuel saving potential of a stove comes from two factors:

- Combustion Efficiency
- Heat Transfer Efficiency

$$\text{Total efficiency} = \text{CE} \times \text{HTE}$$



## The second half of the equation

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Total efficiency = CE x HTE

The simplest of rocket stoves, the insulated elbow alone, can be said to have an overall efficiency of about 18%

If we are getting above 90% CE, what is the HTE?

$18\% = 90\% \times \text{HTE} \quad \text{--} \quad \text{HTE} = 20\%$



## The second half of the equation

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Total Eff = 18% = 90% x 20%

What happens if we work to get CE up to 100% (a difficult 10% increase)?

Total E = 100% x 20% = 20%

But instead what happens if we work to raise HTE by 10% (a much easier increase)?

Total E = 90% x 30% = 27%

**Always work on the weakest link!!!**





## The second half of the equation

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What happens if we work to get CE up to 100% (a difficult 10% increase)?

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But instead what happens if we work to raise HTE by 10% (a much easier increase)?

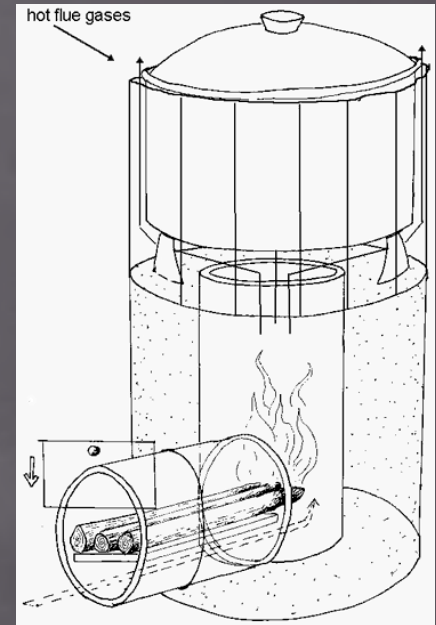
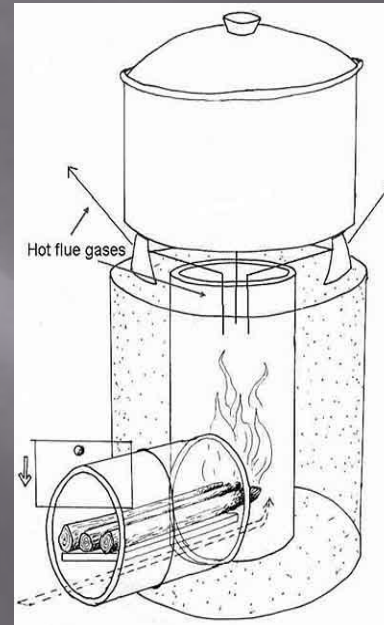
Total E = 90% x 30% = 27%

**Always work on the weakest link!!!**



# Optimising heat transfer

- Maximize radiation to Pot if it does not increase exposure
- Maximize **surface area** of pot that is exposed to hot flue gases
- Keep cross sectional area constant throughout flow path of hot gasses
- Insulate wherever heat is being lost



With a heat exchanger, overall efficiency can be improved by 50% or more



# heat exchanger/skirt

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# Clean burning and fuel efficient stoves



## Complete Combustion

- ▣ Insulated combustion chamber
- ▣ Metered Fuel
- ▣ Metered/Preheated air
- ▣ Pot kept away from Combustion zone

# Clean burning and fuel efficient stoves

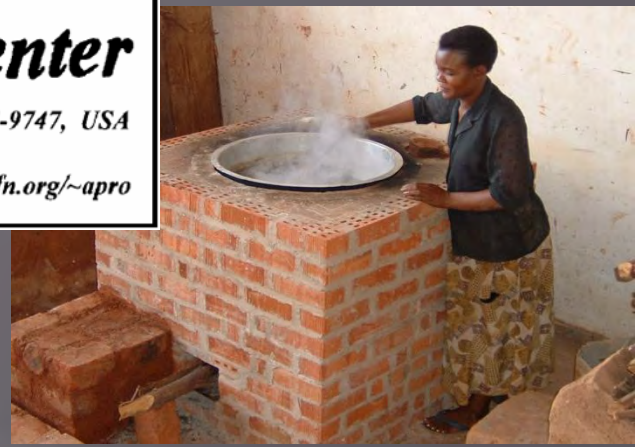


## Improved Heat Transfer

- Radiation if it does not hurt combustion!
- Maximize **surface area**
- Radiation if it does not hurt combustion!
- Constant cross sectional area
- Maximize temp difference between hot gases and pot (insulate against losses)
- Maximize **velocity** of hot flue gases

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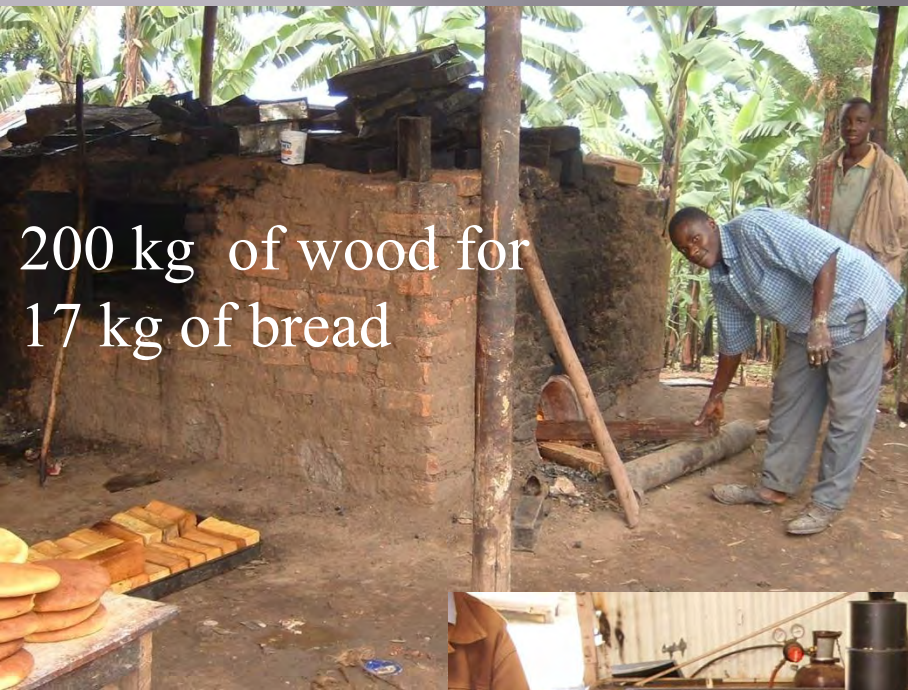
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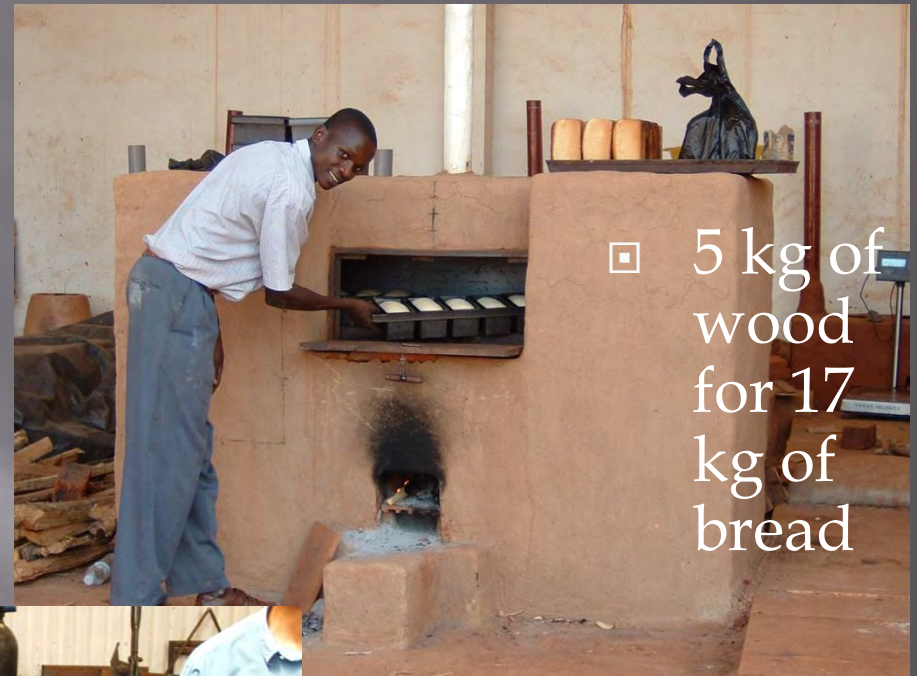
A few rocket stove design possibilities



# Rocket Bread Oven



200 kg of wood for  
17 kg of bread



▣ 5 kg of  
wood  
for 17  
kg of  
bread



08/30/2003

# Central American Griddle Stove

In Central America where tortillas are a major part of the cooking task our griddle stove has been found to save up to 70% or the fuel use





# Tea Estates in Africa



In Southern Africa we have institutional sized rocket stoves at tea plantations that are cooking for 40,000 people



A visual comparison between the quantity of wood used (170kg) for the open fire vs. the amount of wood used (13kg) by the 100L Rocket stove. Independently tested by EP Lauderdale Tea Estates (Malawi)

# Many other stoves



# Many other stoves



*Using the shielded fire stove*

# Many other stoves



*Using the rocket – lorena stove*

# Many other stoves



# Many other stoves



# Many other stoves





# Many other stoves



# Many other stoves



# Many other stoves

